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Quantum Mechanics *P.A.M. Dirac*

Key Concepts, Methods and Machinery

Six postulates in QM

I. On quantum mechanical state

The state of the system is described by the *wavefunction* $\Psi = \Psi(r,t)$, which depends on the coordinates of particle *r* at time t. Wavefunction are in general complex functions of real variables, thus $\,\Psi^*(r,t)$ denotes the complex conjugate of Ψ

$$
P(r,t) = \Psi^*(r,t) \Psi(x,t) dV = |\Psi(r,t)|^2 dV
$$
 (probabilistic interpretation)

II. On operator representation of mechanical quantities

The mechanical quantities that describe the particle (energy, momentum, angular momentum etc.) **are represented by linear operators acting on a** *wavefunction*

The operator of the potential energy

The total energy operator, Hamiltonian: $\overrightarrow{H} = \hat{T} + \hat{V}$

The operator of the kinetic energy

Dirac notation:

$$
\int \!\psi^* \hat{A} \, \phi \! d\, \tau \equiv \! \big\langle \psi \mid \hat{A} \mid \phi \big\rangle
$$

 $\int \psi^* \phi d\,\tau \equiv \bigl\langle \psi \mid \phi \bigr\rangle$

Matrix element of the operator \hat{A}

Scalar product of two wavefunctions

III. On time evolution of the state

The time evolution of the wave function is given by the equation:

$$
\hat{H}\Psi(r,t) = i\hbar \frac{\partial \Psi(r,t)}{\partial t}
$$

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Six postulates in QM

IV. On interpretation of experimental measurements *– not discussed here*

Quantum mechanics in Chemistry

- Let the **molecular system** under study contain atomic nuclei (q_{nuclei}), electrons (*q*electrons) and possibly external fields.
- The key equation in quantum mechanics is the *nonrelativistic* Schrödinger equation:

$$
\hat{H}(\mathbf{q},t)\Psi(\mathbf{q},t) = i\hbar \frac{\partial \Psi(\mathbf{q},t)}{\partial t}
$$

• The vector **q** collects the spatial and spin coordinates of all particles (nuclei and electrons) in the molecular system.

** Postulate III.*

The electronic Schrödinger equation

$$
\hat{H}(\mathbf{q},t)\Psi(\mathbf{q},t) = i\hbar \frac{\partial \Psi(\mathbf{q},t)}{\partial t}
$$

Let the Hamiltonian be time-independent

The electronic Schrödinger equation

The Hamiltonian *(spin-dependent terms not considered)*

$$
\hat{H}_{\text{electrons}} \Psi(\mathbf{q}_{\text{electrons}}) = E \Psi(\mathbf{q}_{\text{electrons}})
$$

through a favorite electronic-structure (quantum-chemical, QC) method.

QC methods are also devised to optimize the spatial configuration of nuclei, leading to E minimization - geometry optimization.

$$
\boxed{\hat{H}_{electrons} \Psi(\mathbf{q}_{electrons}) = E\Psi(\mathbf{q}_{electrons})}
$$
\n
$$
\boxed{\frac{\langle \Psi | \hat{H}_{electrons} | \Psi \rangle}{\langle \Psi | \Psi \rangle}} = E
$$
\n
$$
\boxed{\hat{f} \langle \Psi | \Psi \rangle = 1}
$$
\n
$$
\sum_{i} \langle \Psi | \hat{h}_{one-electron,i} | \Psi \rangle + \sum_{i < j} \langle \Psi | \hat{h}_{two-electron,ij} | \Psi \rangle = E
$$

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The Many Electron Wavefunction

A form for the **electronic wavefunction** that satisfies the permutational antisymmetry (postulate VI) is the **Slater determinant (SD) or a linear combination of SDs.**

Symmetry and spin-adapted **SD** or **linear combination of SDs = configuration state function (CSF)**

$$
\hat{S}^{2}\Psi_{CSF} = S(S+1)\hbar^{2}\Psi_{CSF}
$$

$$
\hat{S}_{z}\Psi_{CSF} = M_{s}\hbar\Psi_{CSF}
$$

$$
\hat{H}_{electrons}\Psi_{CSF} = E\Psi_{CSF}
$$

Molecular orbitals, as a building elements in SD or CSF, are constructed from atomic orbitals:

Hydrogen-like (one-electron) AOs are always of the form:

 $\varphi(r,\theta,\vartheta)$ = $R(r)Y_{lm}(\theta,\vartheta)$ where $R(r)$ is the radial component that decays exponentially with increasing distance from the nucleus $e^{\text{-}\zeta r}$

Since it is impossible to obtain analytic solutions in systems with two or more electrons, the exponential behavior of the AOs – **Slater-type orbitals (STOs)** – were hence the first to be used. They are characterized by an exponential factor in the radial part.

$$
\varphi(r,\theta,\vartheta) = P(r)e^{-\alpha r}Y_{lm}(\theta,\vartheta)
$$

STO

Drawback: difficulties associate with evaluating integrals that appear in the solution of electronic SE.

$$
\varphi(r,\theta,\vartheta) = P(r)e^{-\alpha r^2}Y_{lm}(\theta,\vartheta)
$$

or

$$
\varphi(\alpha,l,m,n;x,y,z) = Ne^{-\alpha r^2}x^l y^m z^n
$$

(Gauss-type orbital GTO)

Drawback: qualitatively incorrect behavior at the nucleus and in the asymptotic limit

Correction

Segmented contraction scheme: each GTO contributes to exactly one CGTO

Balanced basis set - "More art than science"

Minimal basis set - (one STO or GTO or CGTO for one core / valence AO)

DZ

TZ

TZP

TZPD

QZVPD

DZP

Different types of STO/GTO/CGTO functions, e.g.,

polarization functions (P): *e.g., for H atom add p functions for Fe atom add f functions*

diffuse functions (D)

(with small α *in exp(-* α *r²) – allowing to describe electron density at larger distances from nucleus.*

– suitable for anions, soft, large molecules, Rydberg states..

Infinite basis set - *N electrons in ∞ MOs – it requires ∞ AOs* (**ideal but not realistic**)

Effective core potential: *if the core electrons (MOs, AOs) are replaced with an approximate pseudopotential*

double-, triple-, quadruple n-tuple zeta basis sets

DZ, TZ, QZ …

More STO/GTO/CGTO functions describing one AO

General strategies for solving the electronic SE

Family of standard Wave-Function Theories (WFT) – General overview *Welcome to the ZOO*

Two contributions to correlation energy : *static* **and** *dynamic* **correlation**

Hartree–Fock (HF-SCF) method – the Gate to the realm of WFT

Hartree–Fock (HF-SCF) method – Computational Remarks

- *Computational bottleneck – the evaluation of two-electron (four-center) integrals* $\big\langle \varphi_{_P}\varphi_{_r}\,|\, \hat{V}_{_{e-e}}\,|\, \varphi_{_q}\varphi_{_s} \big\rangle$
- *Approximations of such integrals through Cholesky decomposition (CD) or Resolution of Identity (RI-JK).*
	- *Restricted (closed-shell / open-shell HF) unrestricted HF – spin-symmetry broken*

Hartree–Fock (HF-SCF) method – Physical Remarks

- *Each electron experiences the Coulombic repulsion of other electrons through their averaged field (a mean field) (the lack of dynamical correlation – see later)*
- *Exchange interaction among electrons with the same spin orientation (Fermi correlation) – through the antisymmetric nature of the Slater determinant.*
- *One Slater determinant (SD) = one "electronic configuration" ("exact" wavefunction better expressed as a linear combination of many configurations - SDs).*
- *Only the ground-state wavefunction and its energy is solved by HF SCF. (HF not for excited states and their energies)*

Exact non-realistic solution with Full Configuration Interaction (FCI) in the infinite basis set

$$
\begin{array}{c}\n\mathcal{W}_{\infty} \\
\mathcal{W}_{i+4} \\
\mathcal{W}_{i+3}\n\end{array}\n\qquad\n\begin{array}{c}\n\mathcal{H}_{electrons} \Psi_{FCI} = E_{exact} \Psi_{FCI} & \Psi_{FCI} = \sum_{k} C_{k} \Phi_{SD,k} \\
\mathcal{W}_{i+3}\n\end{array}
$$

$$
\langle \Psi_{FCI} | \hat{H}_{electrons} | \Psi_{FCI} \rangle = E_{exact} \qquad (\text{if } \langle \Psi_{FCI} | \Psi_{FCI} \rangle = 1)
$$
\n
$$
\langle \sum_{k} C_{k} \Phi_{k} | \hat{H}_{electrons} | \sum_{l} C_{l} \Phi_{l} \rangle = E_{exact}
$$

Slater-Condon rules \rightarrow many integrals = 0

also Brillouin theorem:

MOs (being orthonormal)

$$
\left\langle \Phi_{\text{HF-SCF}} \mid \hat{\text{H}}_{\text{electrons}} \mid \Phi_i^a \right\rangle = 0
$$

Slater

Exact non-realistic solution with Full Configuration Interaction (FCI) in the infinite basis set

Static *versus* **dynamical** correlation?

dynamical

 $r_{12} \to 0$

Dynamical correlation is related to the Coulomb hole.

• **Short range effects** that arises as **Static ("non-dynamical")**

• from **configurational near-degeneracies** or from deficiencies in Hartree-Fock orbitals

 $\Psi = C_1 \Phi_1 + C_2 \Phi_2$

 $C_1 = 0.7$ $C_2 = 0.3$ *e.g., with*

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Single-reference post-HF approaches (a portion of dynamical correlation included)

Formal scaling behavior of some single-reference QC methods

N – the number of basis functions

Multiconfigurational HF – MCSCF (CASSCF / RASSCF) (a portion of static correlation included)

Modern approaches allowing to extent the active spaces

– Density-matrix renormalization group technology

larger active spaces within **DMRG-CASSCF** (e.g., 30-in-30)

Note on the selection of an active space

"Sometimes trivial, sometimes more difficult, sometimes impossible"

B. Roos

Selection cannot be automatized and depends on the particular system /problem

Chemical insight is important ingredient in choosing a proper active space

Multi-reference wavefunction approaches (a portion of static and dynamic correlation included)

CASPT2 – PT2 on top of CASSCF

Popular for spectroscopy

RASPT2 – PT2 on top of RASSCF

DMRG-CASPT2 – PT2 on top of DMRG-CASSCF Emerging method for "complex electronic structure" chemical transformations

MRCI(SD) – CISD on top of CASSCF

MRCC(SD) – CCSD on top of CASSCF }

Higly accurate but computationally extremely demanding Very small molecules

Density Functional Theory - DFT

To know the Hamiltonian, we need to know the number of electrons and the external potential, i.e. the number of electrons, the nuclear charges, and their positions. All of these can be determined from a knowledge of the density:

$$
\int \rho(r) dr = N \qquad \qquad \frac{\partial}{\partial r_A} \overline{\rho}(r_A)|_{r_A=0} = -2Z_A \overline{\rho}(0)
$$

where $\overline{\rho}$ is the spherical average of the density. The cusps of the *density tell us where the nuclei are.*

^r*(r) therefore determines the Hamiltonian and hence everything about the system.*

Density Functional Theory - DFT

The realm of DFT methods built on two fundamental theorems:

1 st Hohenberg-Kohn theorem:

shows that **electron density of an arbitrary molecular system** (in an electronically nondegenerate ground state) in the absence of external electromagnetic fields **determines unambiguously static external potential** *nuclei* 1

$$
v_{ext}(r) = \sum_{k=1}^{nuclei} Z_k |r - R_k|
$$

2 nd Hohenberg-Kohn theorem:

proves that the **correct ground state electron density minimizes the energy** $E[\rho]$

The total energy is represented as a functional of density:

$$
E[\rho] = V_{ne}[\rho] + T[\rho] + V_{ee}[\rho] = \int \rho(r)v_{ext}(r)dr + T[\rho] + V_{ee}[\rho]
$$

nucleus-electron
attraction energy
of (interacting) electrons
interaction energy

Kohn-Sham Density Functional Theory (KS-DFT)

$$
E[\rho] = \int \rho(r)v(r)dr + T[\rho] + V_{ee}[\rho]
$$
\nCaulomb electron-electron interaction

\n
$$
E[\rho] = \int \rho(r)v_{ex}(r)dr + T_s[\rho] + J[\rho] + (T[\rho] - T_s[\rho]) + (V_{ee}[\rho] - J[\rho])
$$
\nKinetic energy of non-interacting electrons

\n
$$
\frac{1}{2}\int \frac{\rho(r)\rho(r)}{|r-r|}dr dr
$$
\n
$$
E[\rho] = \int \rho(r)v_{ext}(r)dr + T_s[\rho] + J[\rho] + E_{xc}[\rho]
$$
\nExchange-Correlation (XC) Energy

Working Kohn-Sham Equation

The idea of considering the determinantal WF of *N* non-interacting electrons in *N* orbitals, then $T_s[\rho]$ is exactly given as:

$$
T_s[\rho] = \sum_{i=1}^{N} \left\langle \chi_i | -\frac{\hbar^2}{2m_e} \Delta_i | \chi_i \right\rangle
$$
 8. *fulfilling* $\rho = \sum_{i=1}^{N} \left\langle \chi_i | \chi_i \right\rangle$
\nKohn-Sham spinorbital *Real electron density*
\nThen, one-electron KS equation:
\n
$$
\left(-\frac{\hbar^2}{2m_e} \Delta_i + v_{\text{eff}}(r) \right) \chi_i(r) = \varepsilon_i \chi_i(r)
$$
 *(Fock-like
with:* $v_{\text{eff}} = v_{\text{ext}}(r) + \int \frac{\rho(r)}{|r - r|} dr' + v_{\text{xc}}(r)$
\n
$$
LCAO ansatz \Big|
$$
\nAlpha-omega in KS-DFT –
\nexact form unknown

Restricted / Unrestricted Kohn-Sham equations - as in HF $\rho = \rho_a + \rho_\beta$

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Most common of exchange-correlation potentials

- **Local density approximation –** most popular way to do electronic structure calculations in solid state physics
- **Generalized gradient approximation (GGA) –** xc potentials are functionals of electron density and its first spatial derivatives ("gradientcorrected LDA" functionals) *PBE, BP86…*
- **Meta-GGA approximation –** extension of GGA. xc potentials are functionals of electron density, its first and second spatial derivatives and kinetic energy density *TPSS….*
- **Hybrid exchange functionals –** a portion of exact exchange from HF theory is incorporated into xc potentials. Usually, GGA hybrid and GGA approach are combined. *TPSSH, B3LYP, PBE0….*
- **Hybrid exchange and hybrid correlation (double-hybrid) functionals** essentially extension of hybrid-GGA, which uses MP2 correction to replace part of the semi-local GGA correlation. *B2PLYP…*

Limitations of standard KS DFT methods

- **Lack of long-range correlation (dispersion)** empirical corrections $\sim 1/R^6$ *B3LYP+D3*
- **Incorrect long-range exchange behavior**

e.g. incorrect energies of *charge-transfer excitations (exchange should decay asymptotically as r¹² -1 ; B3LYP : 0.2r¹² -1) CAM-B3LYP*

• **Lack of static correlation energy**

Generally lower sensitivity of DFT to multireference character is dependent on the amount of HF exchange included in the functional

• **Self-interaction error**

While the diagonal exchange terms K_{ii} cancel exactly self-interaction Coulomb terms J_{ii} in HF, it is not valid for standard KS-DFT methods. SIE interpreted as the interaction of an electron with itself.

• **Lack of systematic improvability!!!!!**

For a given geometry – wavefunction optimization -> electronic energy *E* **(single-point calculation)**

On the other hand:

QC methods can be also used to optimize geometry – algorithms allowing to evaluate (first, second) derivatives of *E* with respect to the nuclear coordinates and to search crucial points on the potential energy surface

Minima & first-order stationary points (transition states)

(**geometry optimization**)

Thus now, in principle, you are able to read the following sentence:

GGA-type PBE functional in combination with RI-*J* **approximation and the DZP basis set was used for the geometry optimization, while CASPT2(10-in-8) approach combined with a larger basis set (e.g. TZVP) was employed for the final single-point energies.**

APPENDIX

Properties as derivatives of the energy - Bonus

• Consider a molecule in an external electric field ε .

$$
E(\varepsilon) = E(\varepsilon = 0) + \varepsilon \left. \frac{dE}{d\varepsilon} \right|_{\varepsilon = 0} + \frac{1}{2}\varepsilon \left. \frac{d^2 E}{d\varepsilon^2} \right|_{\varepsilon = 0} + \dots
$$

- Dipole moment (μ)
- Polarizability (α)
- First hyperpolarizability (*ß*)

$$
\mu = -\frac{dE}{d\varepsilon}\Big|_{\varepsilon=0}
$$

$$
\alpha = -\frac{d^2E}{d\varepsilon^2}\Big|_{\varepsilon=0}
$$

$$
\beta = -\frac{d^3E}{d\varepsilon^3}\Big|_{\varepsilon=0}
$$

 $dx_i d\varepsilon_\alpha d\varepsilon_\beta$ $dx_i d\varepsilon_\alpha$ $d\varepsilon_{\alpha} d\varepsilon_{\beta} d\varepsilon_{\gamma}$ $d\varepsilon_{\alpha}d\varepsilon_{\beta}$ $d\varepsilon_{\scriptscriptstyle\alpha}^{\,}$ d^3E d^2E $dx_i dx_j dx_k dx_l$ d^4E $dx_i dx_j dx_k$ d^3E $dx_i dx_j$ d^2E *dx i dE* d^3E d^2E *dE dipole moment; in a similar way also multipole moments, electric field gradients, etc. polarizability (first) hyperpolarizability forces on nuclei harmonic force constants; harmonic vibrational frequencies cubic force constants; anharmonic corrections to distances and rotational constants quartic force constants; anharmonic corrections to vibrational frequencies dipole derivatives; infrared intensities polarizability derivatives; Raman intensities*

 $dS_{\alpha}dS_{\beta}$ dS_{α} $dI_{i_\alpha} dJ_{\beta}$ $dB_\alpha dJ_\beta$ $dI_{i_\alpha} dB_\beta$ $dB_{\alpha}dB_{\beta}$ $dI_{i_\alpha}dI_{j_\beta}$ d^2E *dE* d^2E d^2E d^2E d^2E d^2E

 \cdots

magnetizability

nuclear magnetic shielding tensor; relative NMR shifts

indirect spin-spin coupling constants

rotational g-tensor; rotational spectra in magnetic field

nuclear spin-rotation tensor; fine structure in rotational spectra

spin density; hyperfine interaction constants

electronic g-tensor

and many more ...

Restricted Hartree–Fock (RHF) results for LiF

Post-Hartree-Fock for qualitative or quantitative reasons

Why do we want to go beyond the Hartree-Fock description?

- *First, we may wish to improve the accuracy of the computed energy and other properties.*
- *Second, we are dealing with a situation where the Hartree-Fock model is a very poor zeroth-order approximation of the wavefunction.*

Near-degeneracy effects

Ground-state equilibrium properties

Near-degeneracy problems of perturbation theory

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Multireference perturbation theory applied to LiF

Unrestricted UCCSD(T) coupled-cluster calculations of the LiF ground state

- *The UCCSD(T) results compare favorably with the full CI potential energy curve.*
- *The expectation value <S²> is zero for the (unprojected) UHF wavefunction at distances < 3 Å, but* $<$ S² $>$ \approx 1.0 at larger *distances (> 3 Å).*
- *In this example, the spincontamination represents no real problem for the ground state energy.*
- *However, spin-contamination may make the UHF-based methods unsuitable for the study of a variety of molecular properties.*

